

# On the non-steady-state nucleation rate

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# 1-1. Introduction

## cosmic dust universally exists in space

### Milky Way (optical)



### Milky Way (infrared)



Dust grains absorb UV/optical lights and reemit it by their thermal radiation at IR wavelengths!

where did these dust grains come from?

# 1-2. Formation sites of dust

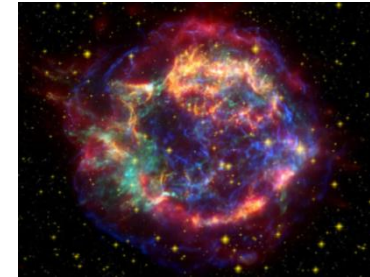
▪ mass-loss winds from AGB stars



▪ expanding ejecta of supernovae (SNe)

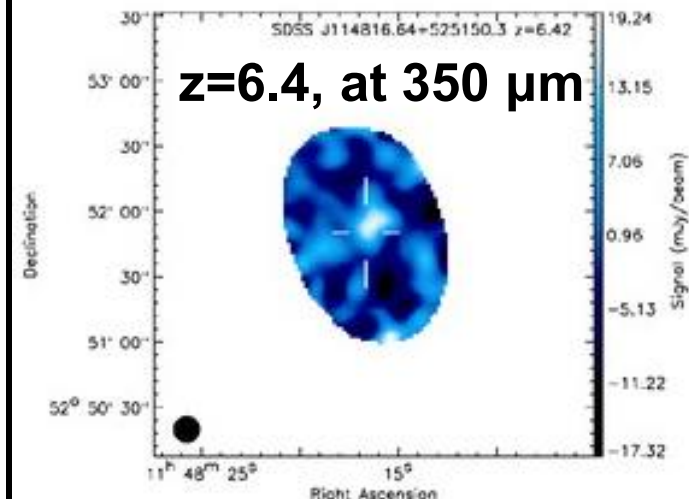
— Huge amounts of dust grains ( $>10^8 M_{\text{sun}}$ ) are detected in quasars at redshift  $z > 5$

→  $0.1 M_{\text{sun}}$  of dust per SN is needed to explain such massive dust at high- $z$

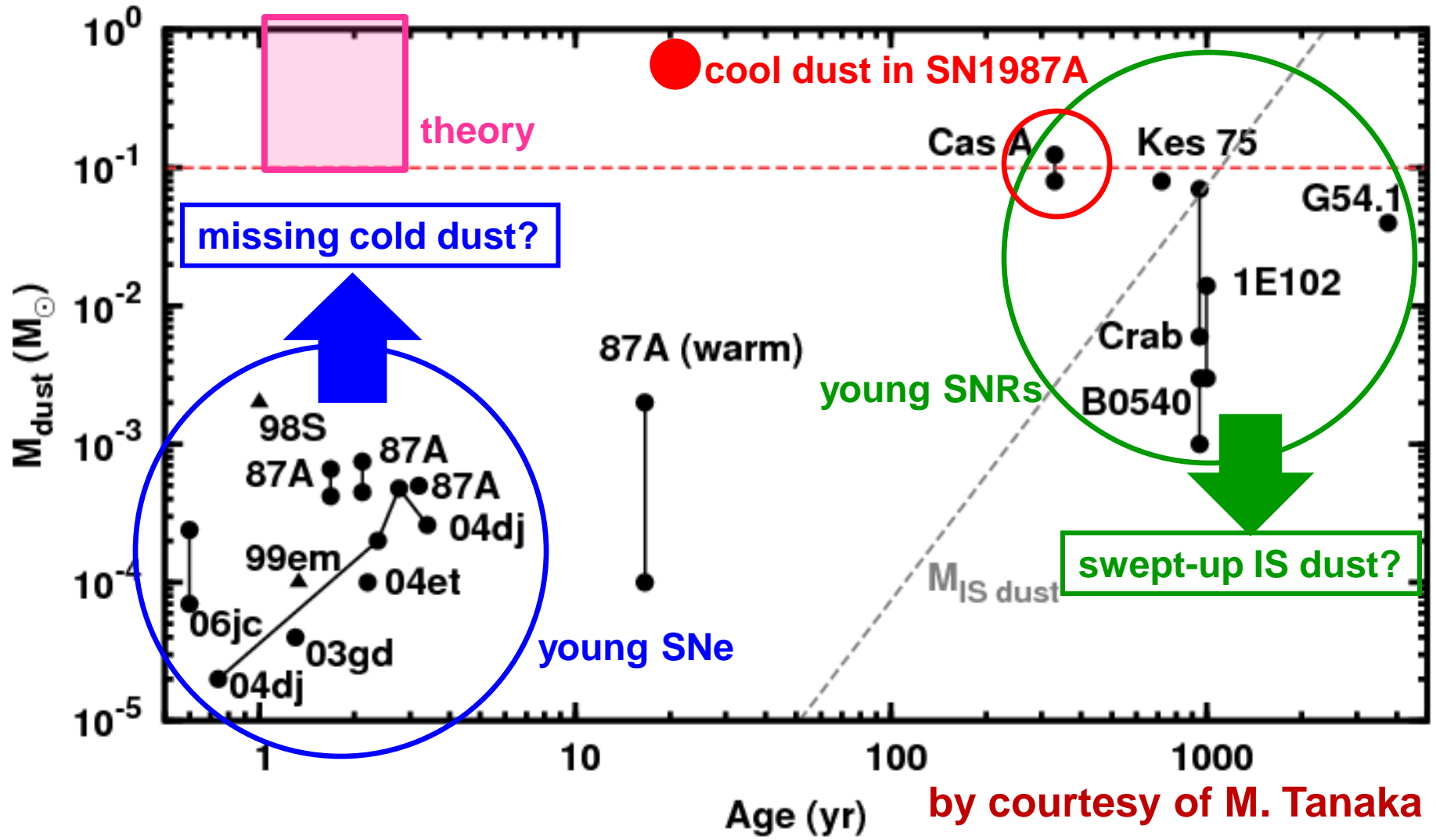


theoretical works predict that 0.1-  
1.0  $M_{\text{sun}}$  of dust can form in SNe  
(e.g., Nozawa et al. 2003)

N/MIR observations detect only  
 $< 10^{-3} M_{\text{sun}}$  of dust in nearby SNe  
(e.g., Sakon et al. 2009)



# 1-3. Summary of dust mass in CC-SNe



by courtesy of M. Tanaka

**Far-IR to sub-mm observations are essential for revealing the mass of dust grains produced in the ejecta of SNe**

## 2-1. Supersaturation ratio

### • supersaturation ratio, S

→ ratio of partial pressure  $P_1$  to equilibrium partial pressure  $P^0_1$

$$\ln S = \ln \left( \frac{p_1}{\dot{p}_1} \right) = -\frac{1}{kT} (\dot{g}_s - \dot{g}_1) + \ln \left( \frac{p_1}{p_0} \right)$$

For condensation of dust,

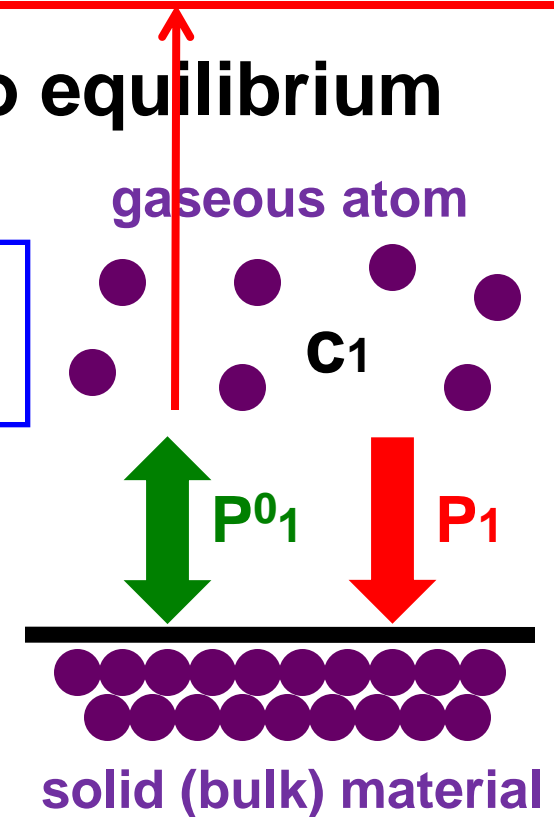
$$S = P_1 / P^0_1 > 1$$

$$\rightarrow \ln S = \ln(P_1 / P^0_1) > 0$$

$$\ln S = \frac{A}{T} - B + \ln \left( \frac{c_1 kT}{p_0} \right)$$

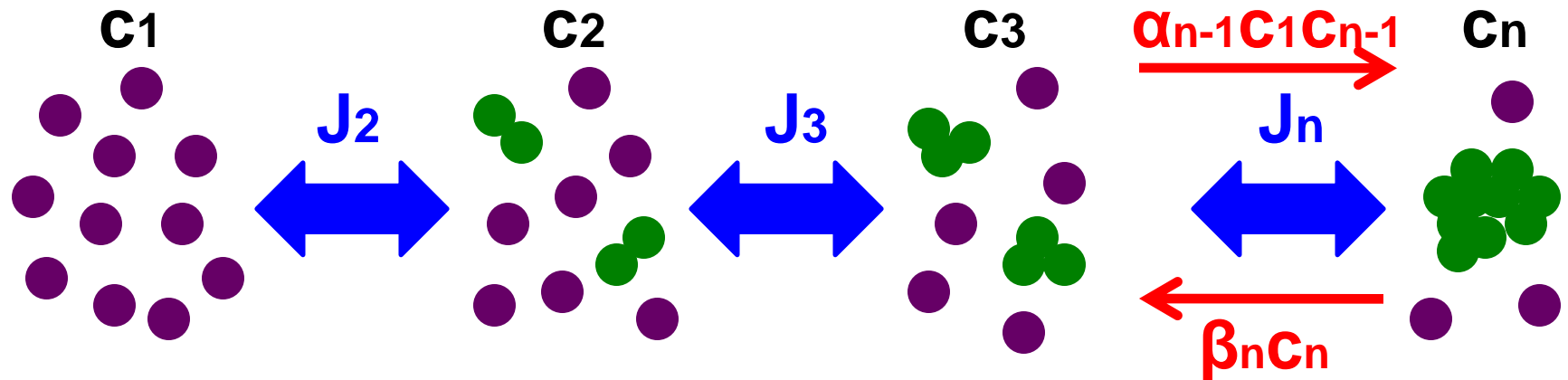
where  $-\Delta G^0/kT = A/T - B$   
( $A=8.64 \times 10^4$ ,  $B=19$  for C grain)

$T_{\text{dust}}$  is assumed to equal to  $T=T_{\text{gas}}$



$\ln S$  is higher for lower  $T$  and higher  $c_1$

## 2-2. Concept of nucleation theory



### ▪ master equations

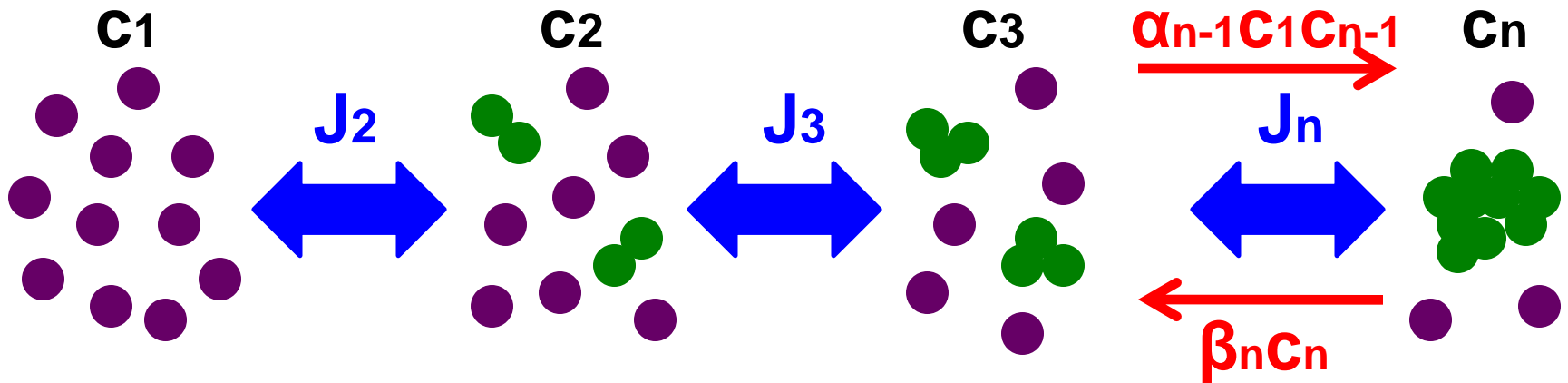
$$\frac{dc_n}{dt} = J_n(t) - J_{n+1}(t) \quad \text{for } 2 \leq n \leq n_*,$$

$$J_n(t) = \alpha_{n-1} c_{n-1} c_1 - \beta_n c_n \quad \text{for } 2 \leq n \leq n_*,$$

$$\alpha_n = \frac{s_n}{1 + \delta_{1n}} 4\pi a_0^2 n^{\frac{2}{3}} \left( \frac{kT}{2\pi m_n} \right)^{\frac{1}{2}},$$

$$\beta_n = \alpha_{n-1} \frac{\overset{\circ}{c}_{n-1}}{\overset{\circ}{c}_n} \overset{\circ}{c}_1,$$

## 2-3. Steady-state nucleation rate



▪ current density  $J_n$

$$J_n(t) = \alpha_{n-1} c_1 \left( c_{n-1} - c_n \frac{\dot{c}_{n-1}}{\dot{c}_n} \frac{\dot{c}_1}{c_1} \right).$$

▪ “steady”  $\rightarrow J_s = J_2 = J_3 = \dots = J_\infty$

$$\frac{1}{J_s} = \frac{1}{\alpha_1 c_1^2} + \sum_{i=2}^{\infty} \frac{1}{\alpha_i c_1^2} \exp(\gamma'_i).$$

# 2-4. Critical number of atom for nucleation

$$\frac{1}{J_s} = \frac{1}{\alpha_1 c_1^2} + \sum_{i=2}^{\infty} \frac{1}{\alpha_i c_1^2} \exp(\gamma'_i).$$

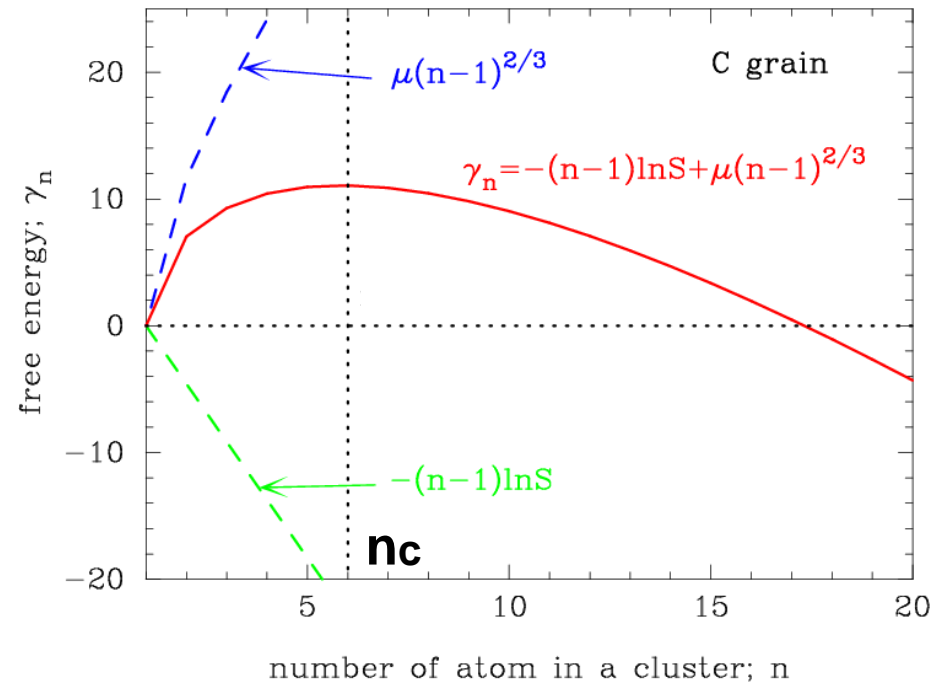
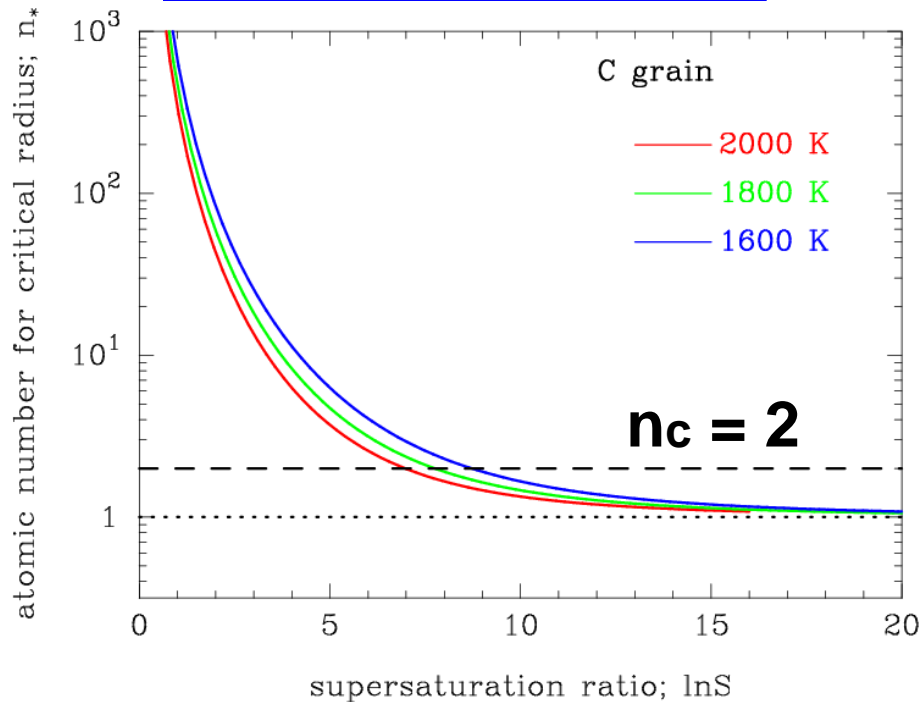
$$\gamma'_n = \mu (n-1)^{2/3} - (n-1) \ln S,$$

critical number:  $n_c$

$$(n_c - 1)^{1/3} = \frac{2}{3} \frac{\mu}{\ln S}.$$

$$\mu = 4\pi a_0^2 \sigma / kT$$

( $\sigma = 1400 \text{ erg cm}^{-2}$  for bulk C materials)





# 2-5. Non-steady-state nucleation

## ▪ steady-state nucleation rate: $J_s$

$$J_s = s \Omega_0 \left( \frac{2\sigma}{\pi m_1} \right)^{\frac{1}{2}} c_1^2 \exp \left[ -\frac{4}{27} \frac{\mu^3}{(\ln S)^2} \right].$$

$$J_n(t) = \alpha_{n-1} c_1 \left( c_{n-1} - c_n \frac{\dot{c}_{n-1}}{\dot{c}_n} \frac{\dot{c}_1}{c_1} \right).$$



$$J_n(t) = \alpha_{n-1} c_1 [c_{n-1} - c_n \exp(\gamma_n)]$$

↑

→  $J_s = J_2 = J_3 = \dots = J_\infty$

$$\gamma'_n = \mu (n-1)^{\frac{3}{2}} - (n-1) \ln S,$$

$$\gamma_n = \mu \left[ (n-1)^{\frac{3}{2}} - (n-2)^{\frac{3}{2}} \right] - \ln S,$$

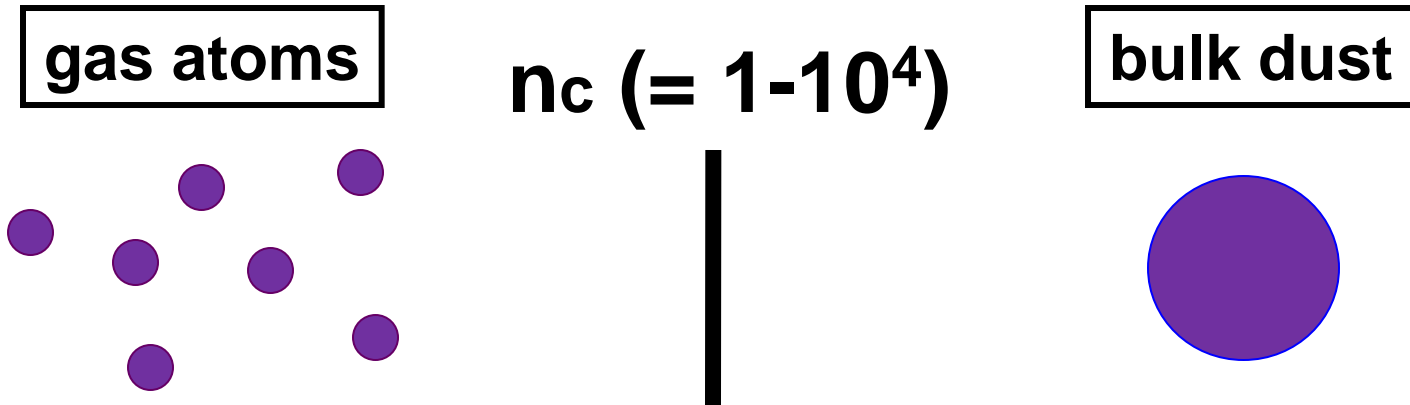
## ▪ non-steady-state nucleation

$n^* = 100$

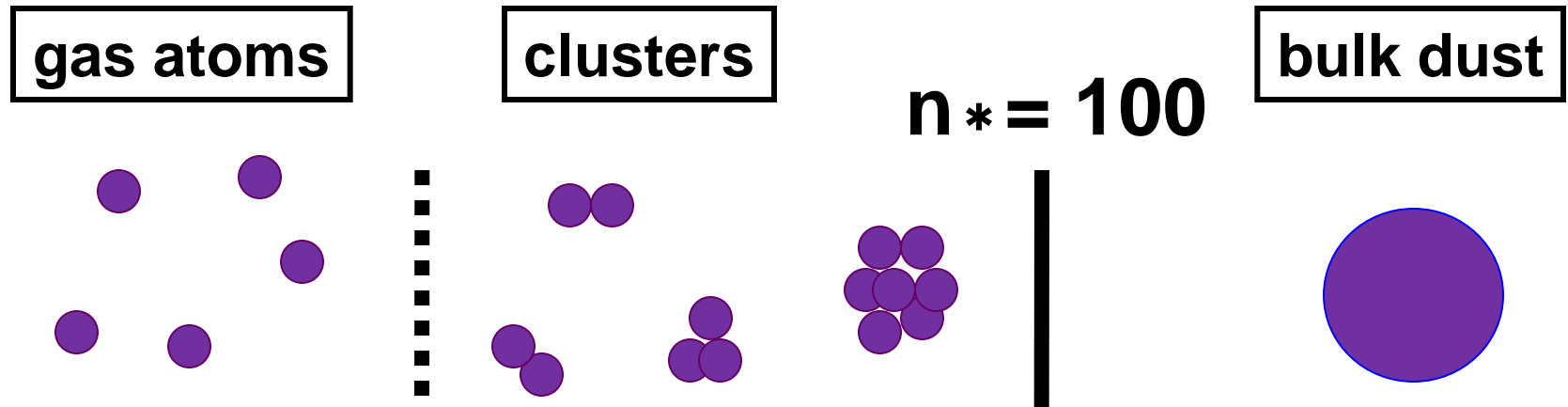
$$\frac{dc_n}{dt} = J_n(t) - J_{n+1}(t) \quad \text{for } 2 \leq n \leq n^*,$$

## 2-6. Steady and non-steady

### ▪ steady-state nucleation rate: $J_s$



### ▪ non-steady-state nucleation rate: $J^*$



# 3-1. Basic equations for dust formation

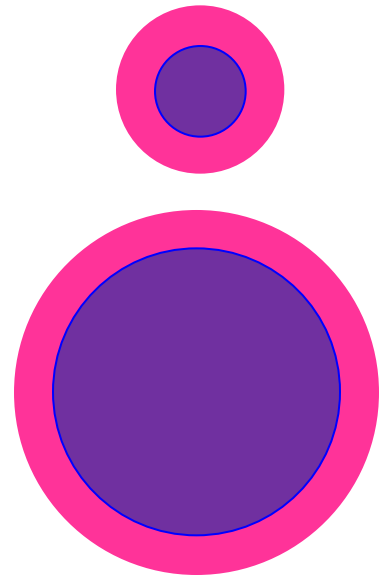
## Equation of mass conservation

$$c_{10} - c_1 = \underbrace{\sum_{n=2}^{n_*-1} n c_n}_{\text{(cluster)}} + \underbrace{\int_{t_0}^t J_{n_*}(t') \frac{a^3(t, t')}{a_0^3} dt'}_{\text{(bulk dust)}}$$

## Equation of grain growth

$$\frac{da}{dt} = s\Omega_0 \left( \frac{kT}{2\pi m_1} \right)^{\frac{1}{2}} c_1 \left( 1 - \frac{1}{S} \right),$$

$$\frac{dV}{dt} = s\Omega_0 4\pi a^2 \left( \frac{kT}{2\pi m_1} \right)^{\frac{1}{2}} c_1 \left( 1 - \frac{1}{S} \right),$$



Growth rate is independent of grain radius

# 3-2. Evolution of gas temperature and density

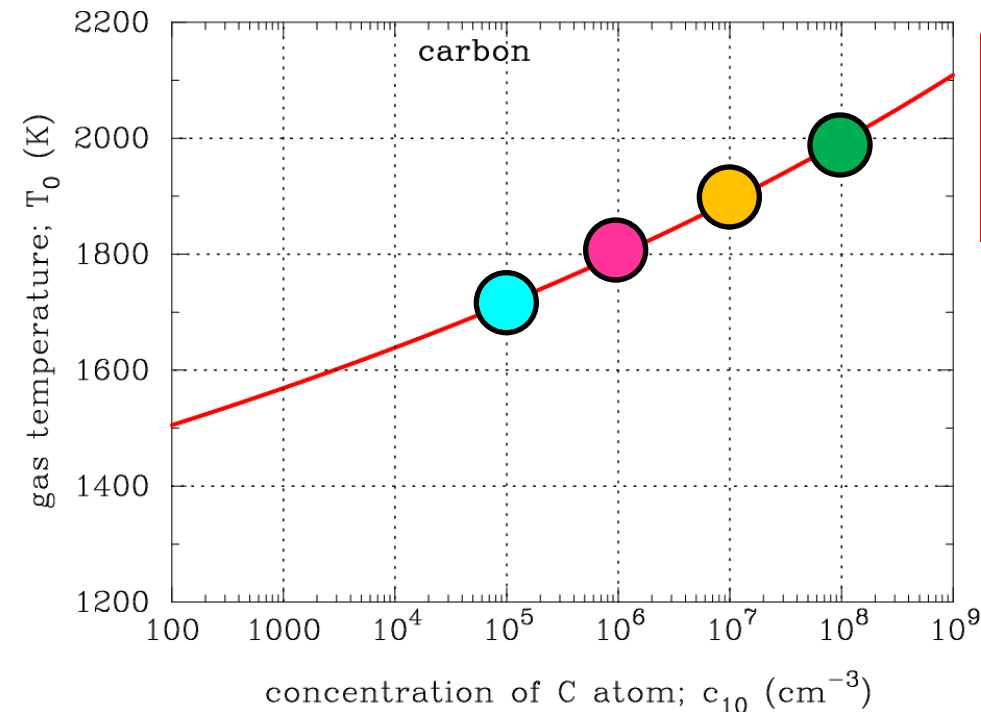
## Time evolution of $c_1$ and $T$

$$c_1(t) = c_{10} \exp\left(-\frac{t - t_0}{\tau_{\text{exp}}}\right)$$

$$T(t) = T_0 \exp\left(-\frac{t - t_0}{\tau_{\text{cool}}}\right),$$

$\tau_{\text{exp}} = 100$  day

$\tau_{\text{cool}} = 100$  day



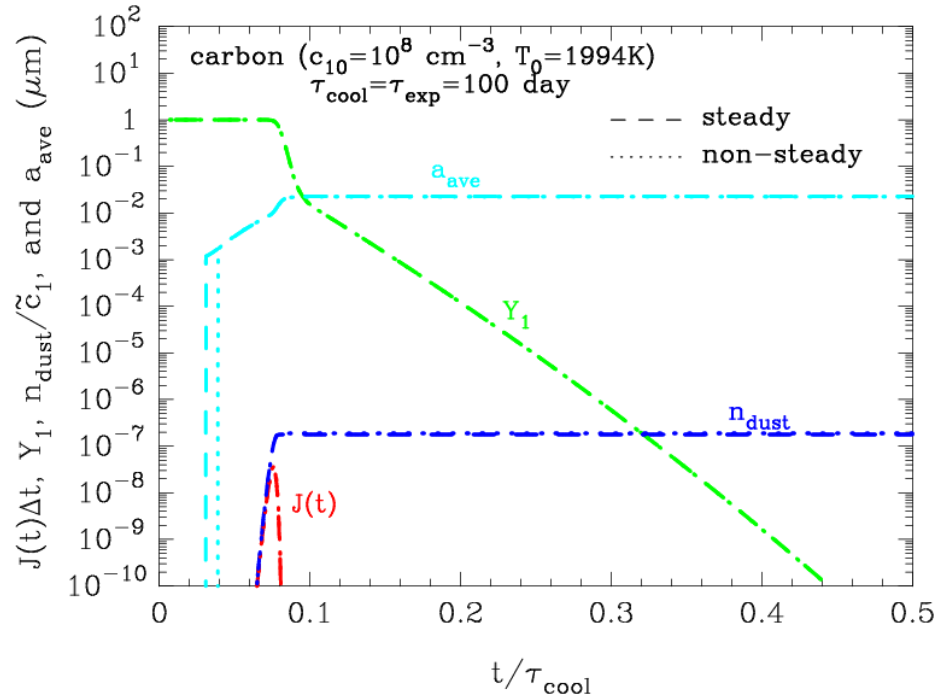
$$\ln S = \frac{A}{T_0} - B + \ln\left(\frac{c_{10} k T_0}{p_0}\right) = 0$$

$$c_{10} = 10^8, 10^7, 10^6, 10^5 \text{ cm}^{-3}$$

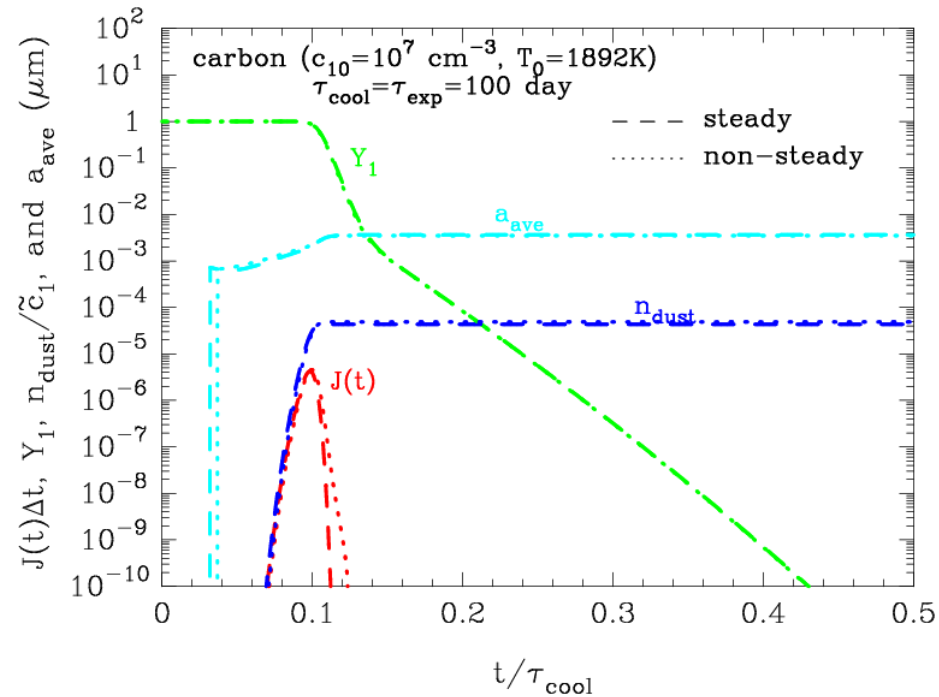
$$S = 1.0$$

# 4-1. Steady vs. Non-steady (1)

$c_{10} = 10^8 \text{ cm}^{-3}$



$c_{10} = 10^7 \text{ cm}^{-3}$

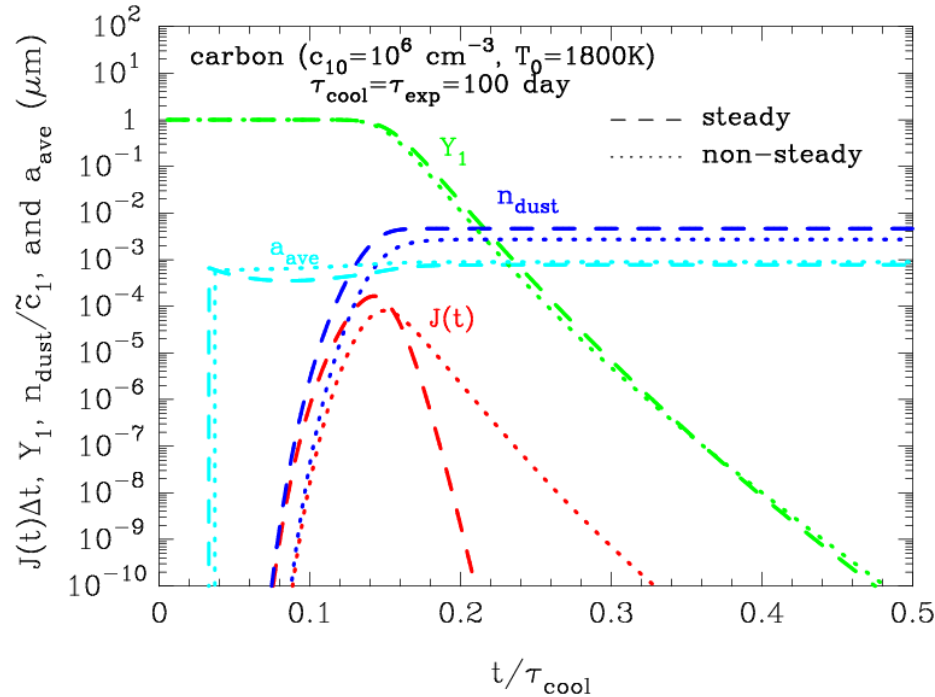


- dashed line : steady-state nucleation
- dotted line : non-steady-state nucleation

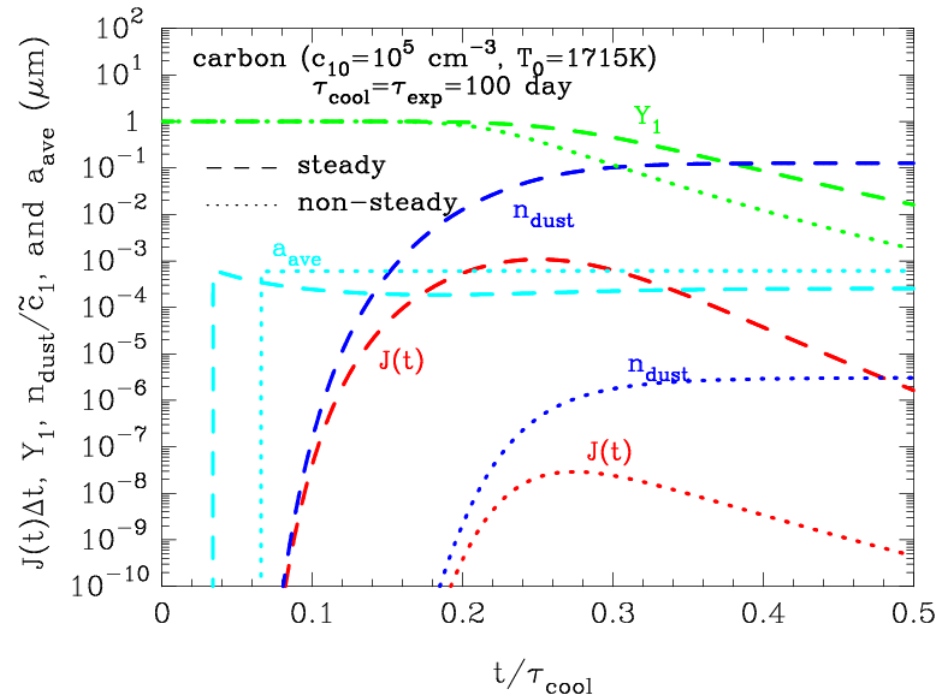
The difference between steady and non-steady nucleation is small for higher initial densities

# 4-2. Steady vs. Non-steady (2)

$c_{10} = 10^6 \text{ cm}^{-3}$



$c_{10} = 10^5 \text{ cm}^{-3}$



- dashed line : steady-state nucleation
- dotted line : non-steady-state nucleation

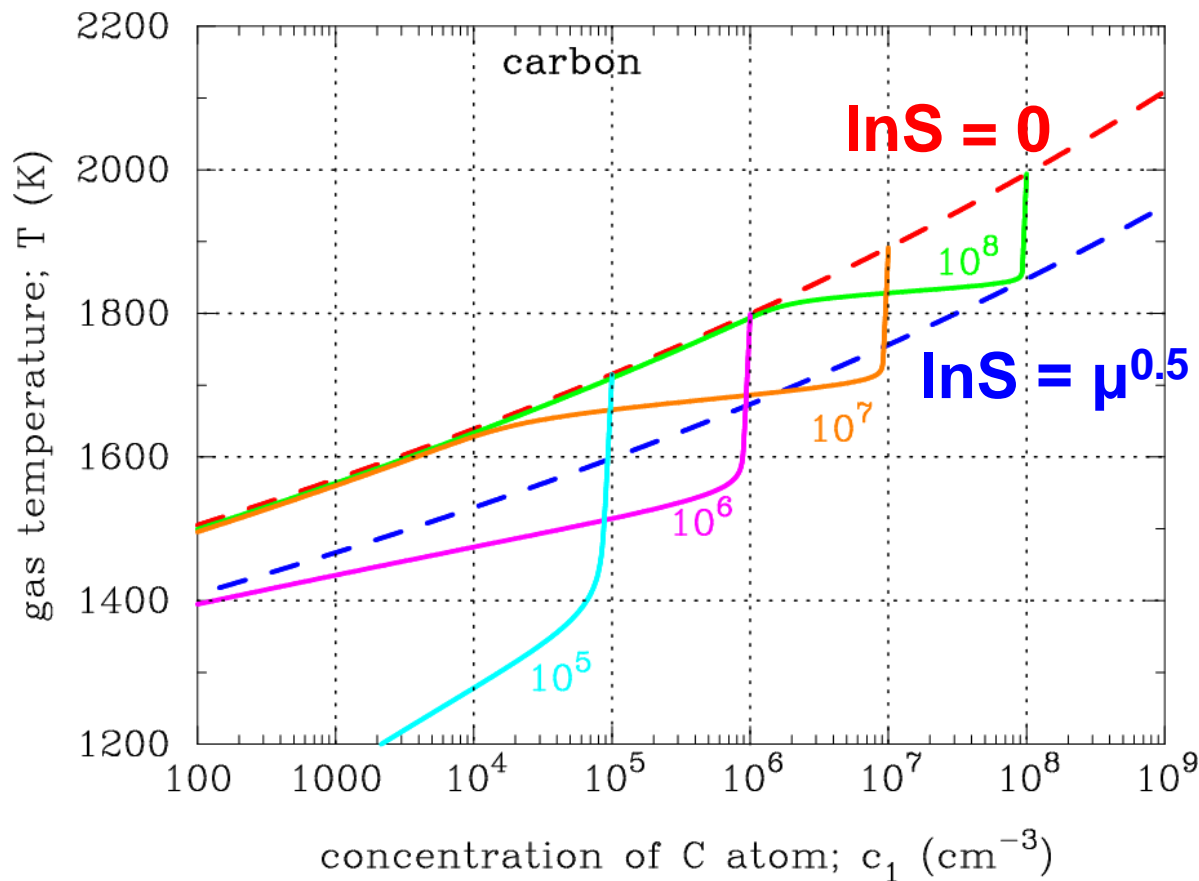
The difference between steady and non-steady seems significant for lower densities

# 4-3. Relaxation time

• relaxation time,  $T_{\text{relax}}$  (e.g., Gail et al. 1984)

$$\rightarrow T_{\text{relax}} = T_{\text{coll}} \left[ \frac{(\ln S)^2}{\mu} \right]$$

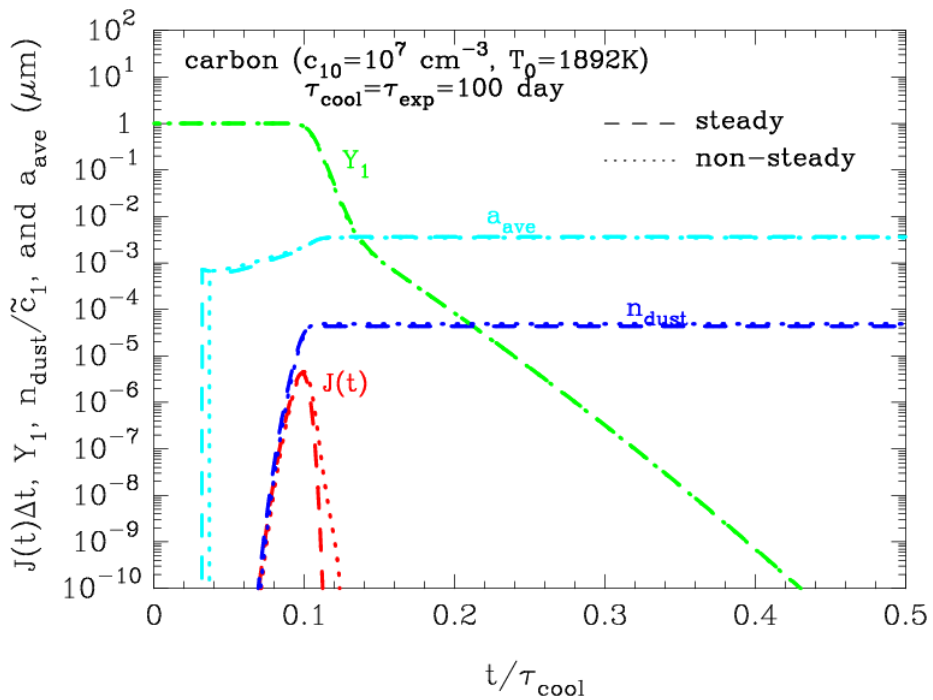
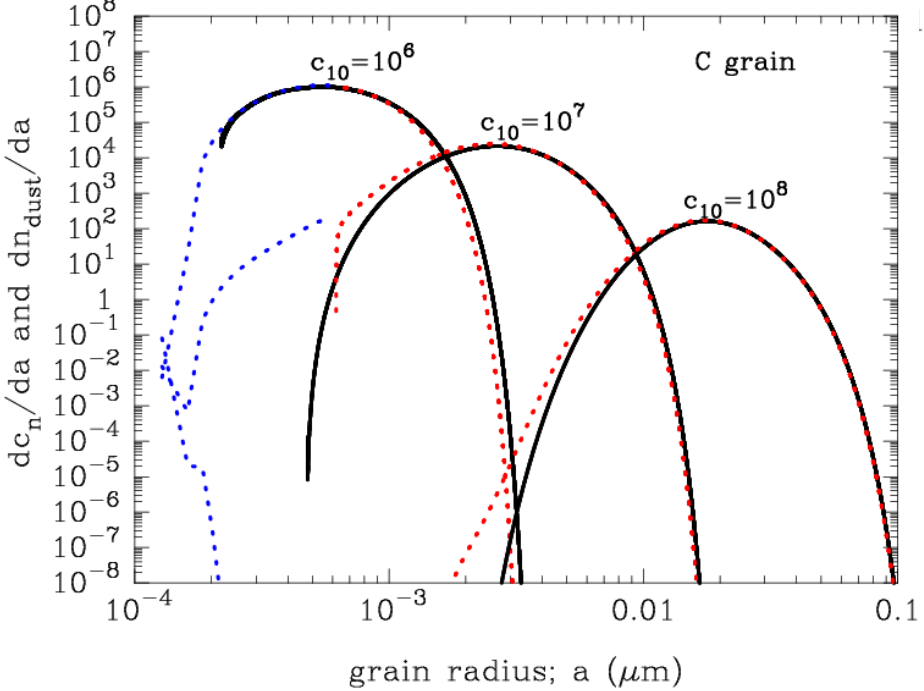
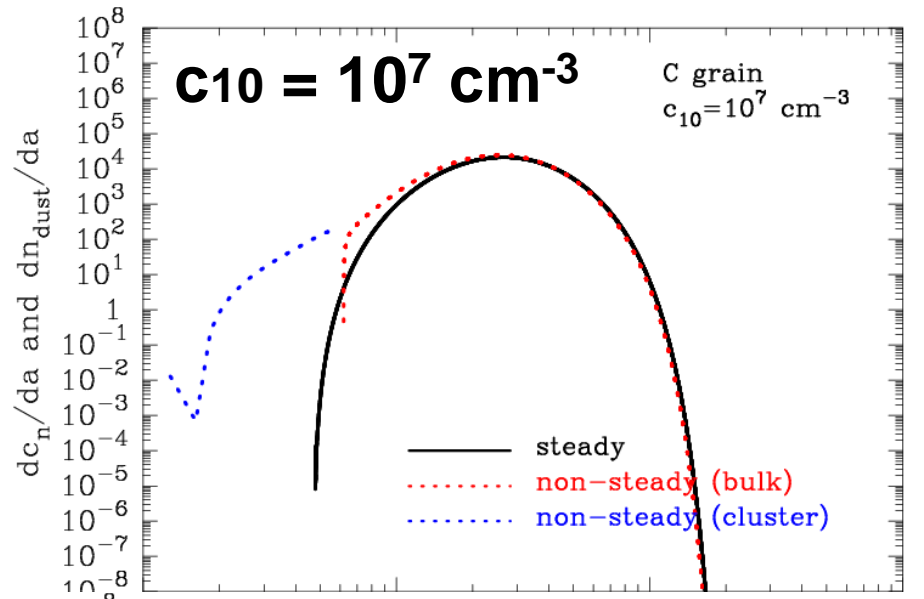
$$T_{\text{coll}} = \left[ S_n 4\pi a_0^2 \langle v \rangle c_1 \right]^{-1}$$



$T_{\text{relax}} / T_{\text{coll}} =$   
 $\left[ \frac{(\ln S)^2}{\mu} \right] < 1$   
 $\rightarrow$  steady

$T_{\text{relax}} / T_{\text{coll}} =$   
 $\left[ \frac{(\ln S)^2}{\mu} \right] > 1$   
 $\rightarrow$  non-steady

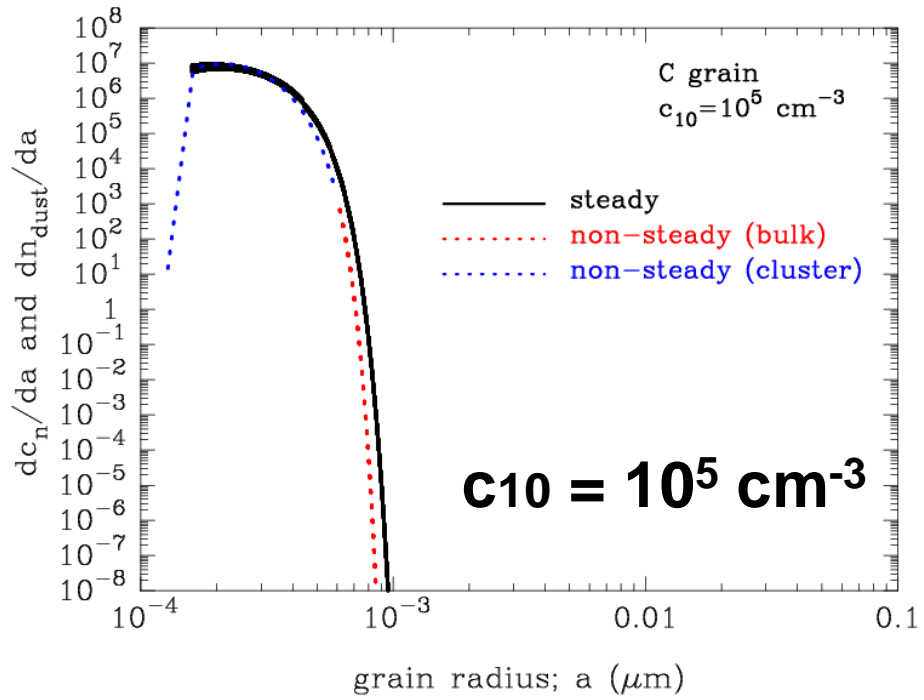
# 5-1. Steady vs. Non-steady: size distribution (1)



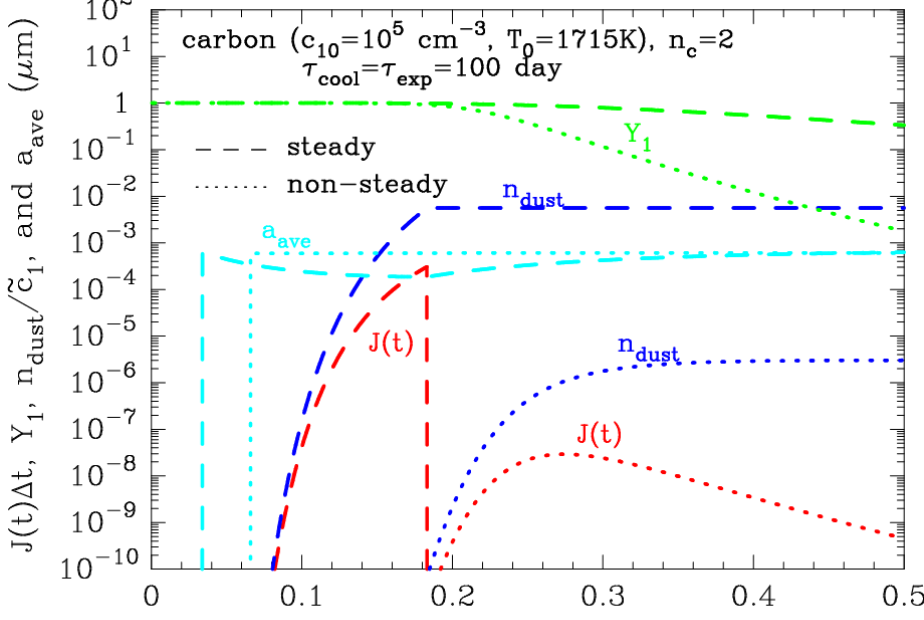
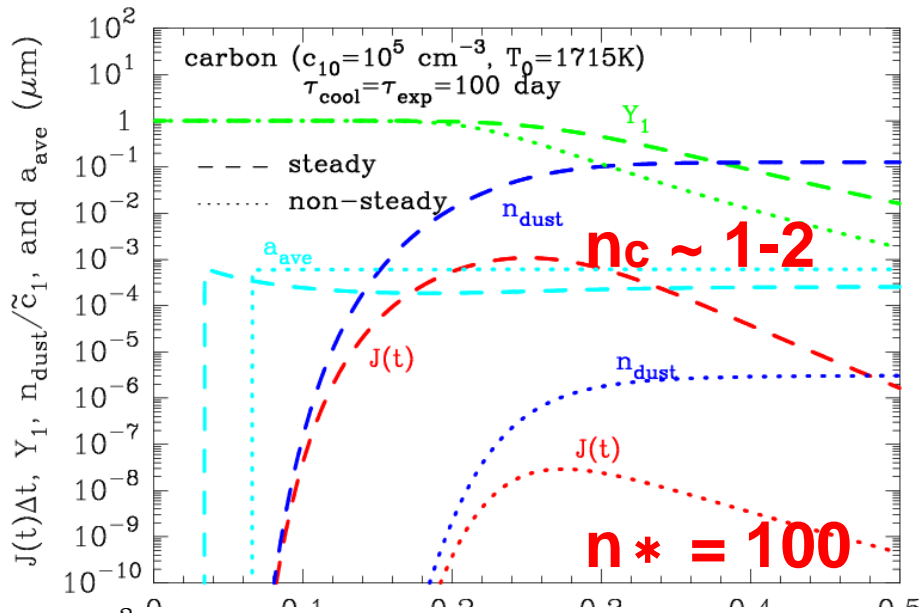
**The size distribution of bulk grains from non-steady-state is consistent with that from steady-state**



# 5-2. Steady vs. Non-steady: size distribution (2)

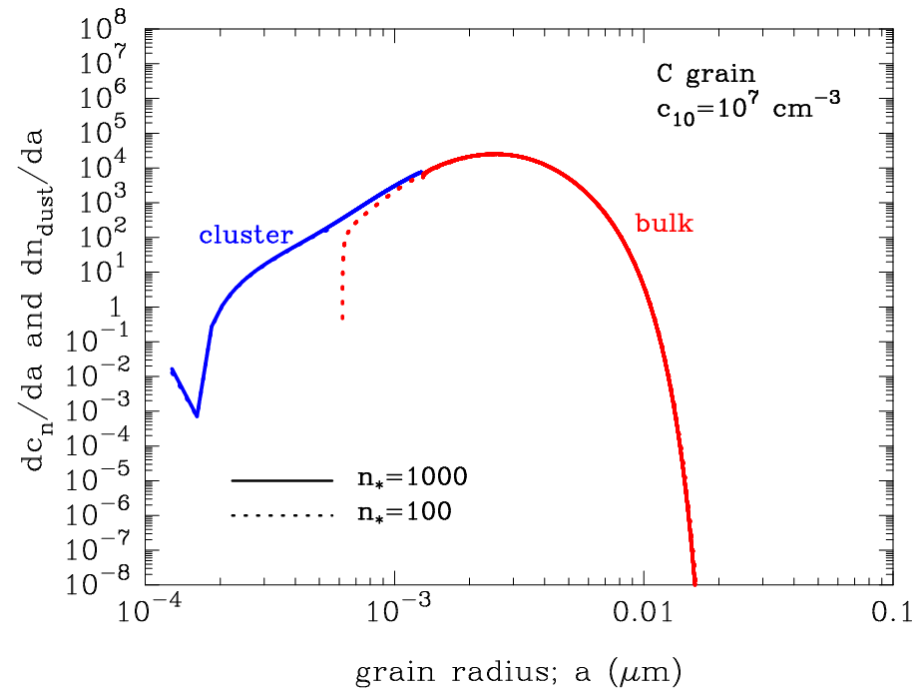
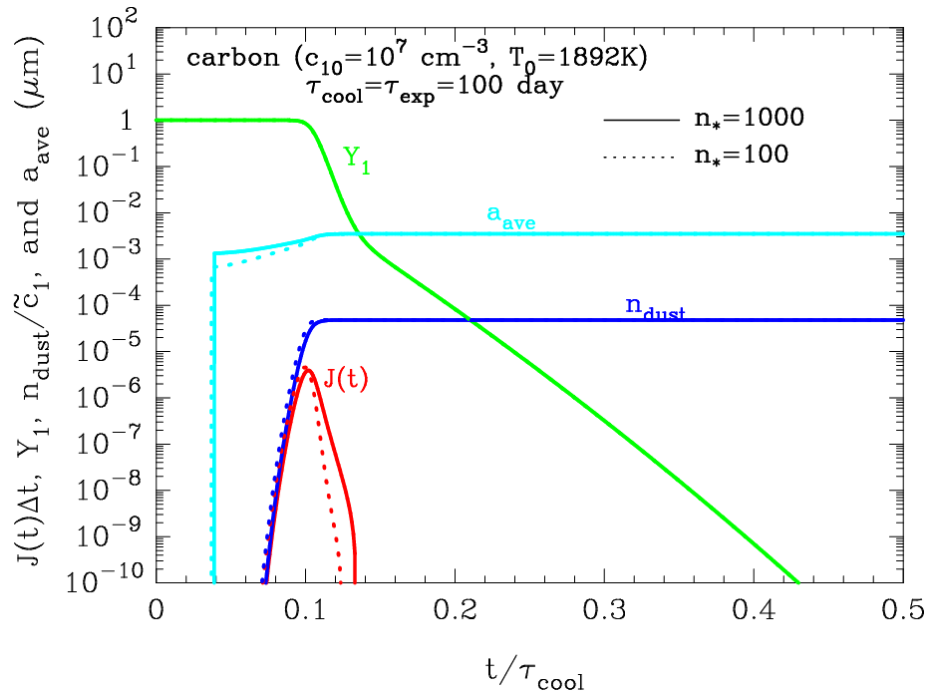


**Even for  $c_{10} = 10^5 \text{ cm}^{-3}$ , the size distribution is quite similar between steady-state and non-steady-state**



$t/\tau_{cool}$

# 5-3. Dependence on cluster maximum size

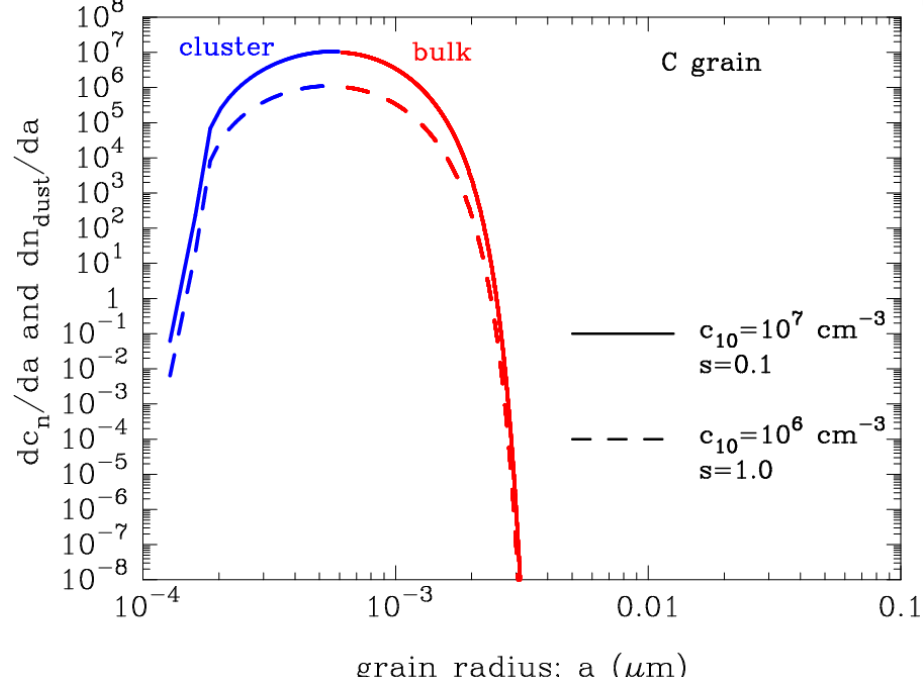
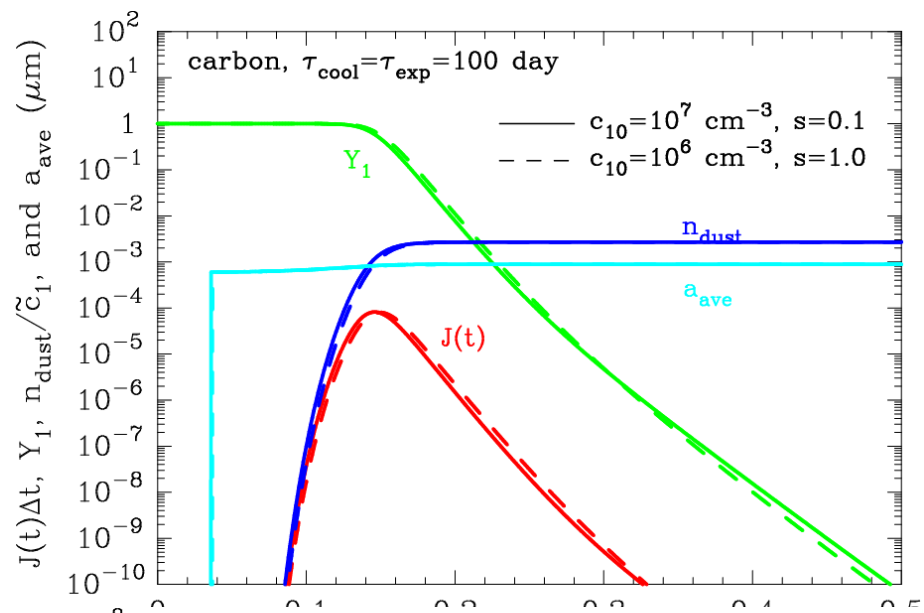
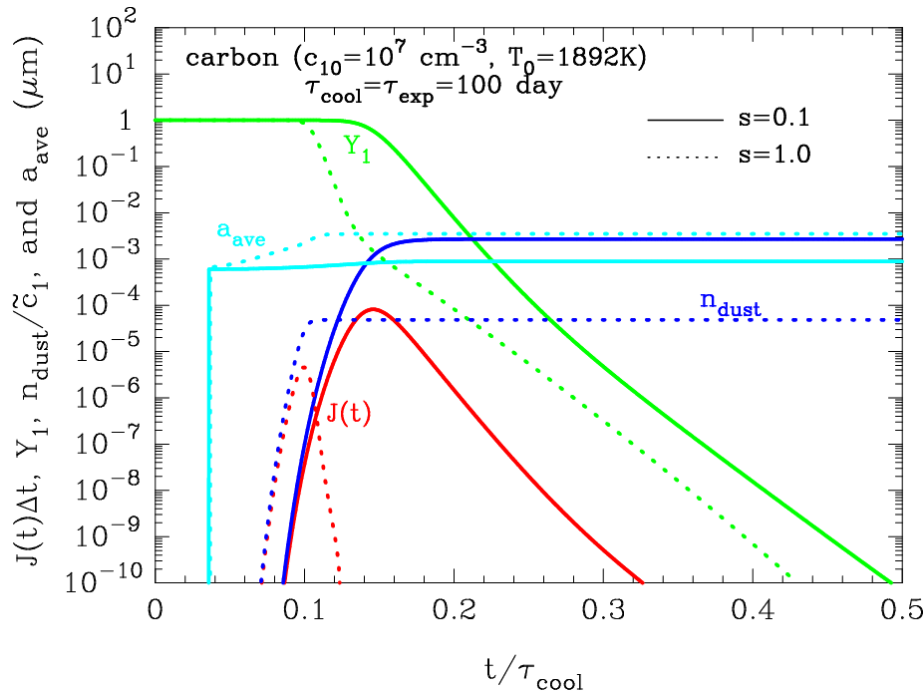


- solid line : non-steady-state,  $n_* = 1000$
- dotted line : non-steady-state,  $n_* = 100$

**The size distribution does not depend on  $n_*$**

**How many atoms are needed to be defined as bulk dust grains?**

# 5-4. Dependence on sticking probability



**The result for  $c_{10} = 10^7 \text{ cm}^{-3}$  and  $s = 0.1$  is similar to that for  $c_{10} = 10^6 \text{ cm}^{-3}$  and  $s = 1$ , but not completely same**

# 6. Summary

- The difference between steady and non-steady state nucleation is not significant
  - steady-state nucleation rate is a good approximate
  - what is the definition of bulk dust grains?
- In the future work
  - to extend to multiple-element grains like silicate
  - to calculate the temperature of dust (clusters)
    - Dust (cluster) temperature is assumed to be equivalent to gas temperature
  - unknown quantities: sticking probability, shape and surface energy of small clusters